

# **Well Systems – An Introduction to Operation and Maintenance**

**Maryland Center for Environmental Training**  
**301-934-7500**  
**[info@mcet.org](mailto:info@mcet.org)**  
**[www.mcet.org](http://www.mcet.org)**

Well Systems:  
An Introduction to Operation and  
Maintenance

Eddie Cope, CET

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Introductions

- Instructor
- Course Participants
  - Name
  - Where you work
  - Type of system(s) that you operate
  - Expectations of this training

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Topic to be covered

- Operator Responsibilities
- Regulations
- Water Use and the Water Cycle (Hydrology)
- Groundwater Sources and Aquifers
- Wells and Wellhead Protection
- Well Permitting and Construction
- Well Operation and Maintenance

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**Topics continued**

- Well Rehabilitation
- Well Sampling and Testing
- Contaminants and Treatment

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**Regulatory  
Setting**

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**Public Water Supply  
Regulations**

- The Safe Drinking Water Act of 1974

Directs the EPA to establish standards and requirements necessary to protect the public from all known harmful contaminants in drinking water.



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## Classification by Treatment Process (MDE)

Class of Plants	Type of Treatment Systems	Typical Processes Included in the System
1	Disinfection	Chlorination
2	Chemical Treatment	Chlorination, pH control Fluoridation
3	Simple Iron Removal	Chlorination, pH control, fluoridation, filtration, and iron removal utilizing ion exchange or contact oxidation processes.
4	Complete Treatment	Chlorination, pH control, fluoridation, aeration, coagulation, sedimentation, filtration, and complex iron removal.
5	Site Specific	Site specific - any alternative technology plant not covered under the classification system.
6	No Chemical Treatment	Well, storage tanks, UV disinfection

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## Well Permitting

**MARYLAND DEPARTMENT OF THE ENVIRONMENT**  
 WATER MANAGEMENT ADMINISTRATION - Water Supply Program  
 1102 Montross Building • Baltimore, Maryland 21201  
 (410) 537-2000 • (410) 535-4101 • Fax (410) 537-2014 • E-mail: wma@state.md.us

**APPLICATION TO APPROPRIATE AND USE WATERS OF THE STATE**  
 Change of Existing Permit

① New Application      ② Renew Application

**APPLICATION**  
 Name: Harold County - SW - Water Operations      1812220111  
 City/Town: None      None  
 Address: 1311 Harwell Pike Road      Millersville MD 21114

**WITNESS OF GROUNDWATER**      **WITNESS OF SURFACE WATER**  
 I, None      I, None  
 certify that the above information is true and correct to the best of my knowledge and belief.

**PROPOSED GROUNDWATER**      **PROPOSED SURFACE WATER**  
 Estimated annual withdrawal: None      Estimated annual withdrawal: None  
 Estimated daily withdrawal: None      Estimated daily withdrawal: None  
 Estimated peak withdrawal: None      Estimated peak withdrawal: None  
 Estimated withdrawal during peak flow: None      Estimated withdrawal during peak flow: None  
 Estimated withdrawal during low flow: None      Estimated withdrawal during low flow: None  
 Estimated withdrawal during dry weather: None      Estimated withdrawal during dry weather: None  
 Estimated withdrawal during wet weather: None      Estimated withdrawal during wet weather: None  
 Estimated withdrawal during flood: None      Estimated withdrawal during flood: None  
 Estimated withdrawal during drought: None      Estimated withdrawal during drought: None  
 Estimated withdrawal during other: None      Estimated withdrawal during other: None

**FACTORY DESIGN**  
 Design: None      None  
 Capacity: None      None  
 Location: None      None  
 Construction: None      None  
 Operation: None      None  
 Maintenance: None      None  
 Other: None      None

**PURPOSE**  
 A. Drinking Water      B. Industrial Process Water      C. Other      D. Other      E. Other      F. Other      G. Other      H. Other      I. Other      J. Other      K. Other      L. Other      M. Other      N. Other      O. Other      P. Other      Q. Other      R. Other      S. Other      T. Other      U. Other      V. Other      W. Other      X. Other      Y. Other      Z. Other      AA. Other      AB. Other      AC. Other      AD. Other      AE. Other      AF. Other      AG. Other      AH. Other      AI. Other      AJ. Other      AK. Other      AL. Other      AM. Other      AN. Other      AO. Other      AP. Other      AQ. Other      AR. Other      AS. Other      AT. Other      AU. Other      AV. Other      AW. Other      AX. Other      AY. Other      AZ. Other      BA. Other      BB. Other      BC. Other      BD. Other      BE. Other      BF. Other      BG. Other      BH. Other      BI. Other      BJ. Other      BK. Other      BL. Other      BM. Other      BN. Other      BO. 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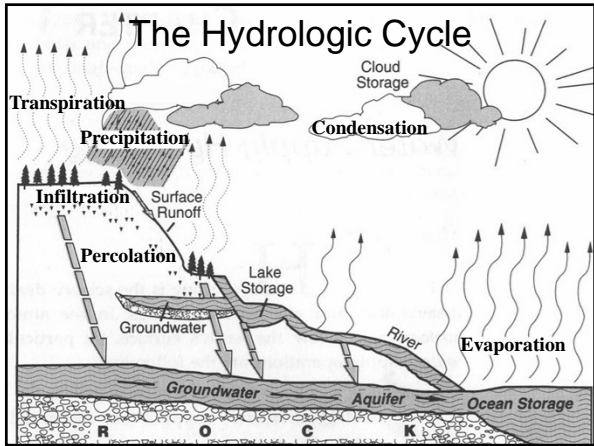
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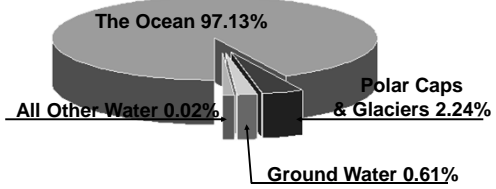
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## Where is the Water?



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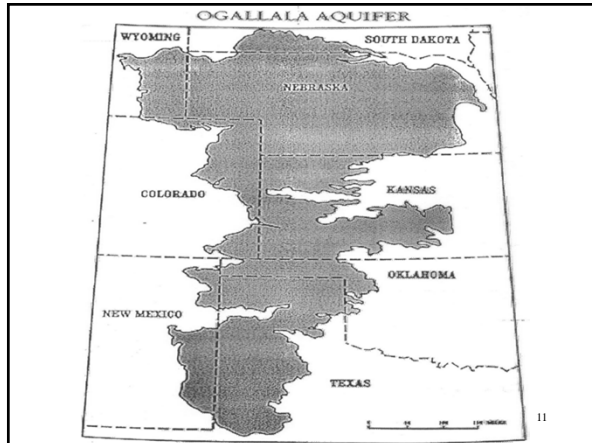
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## Groundwater Sources

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## Important Terms

- Aquifer
- Static Water Level
- Pumping Water Level
- Drawdown
- Residual Drawdown
- Well Yield
- Specific capacity
- Transmissivity

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## Important Terms (cont.)

- Permeability
- Porosity
- Mutual Interference

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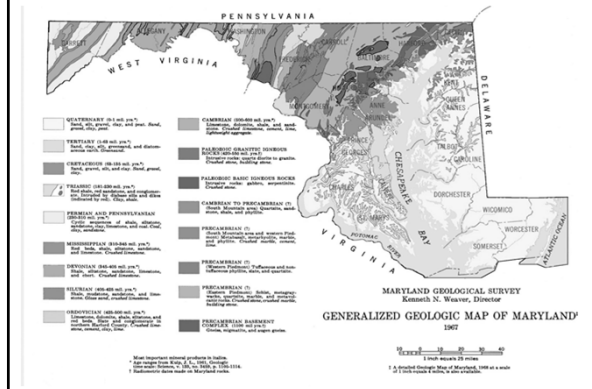
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## Generalized geologic map of MD




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## MD Geology

- 3 regions: Coastal Plain, Piedmont, and Blue Valley and Ridge and Appalachian Plateau.
- Coastal Plain: gravel, sand, silt and clay
- Piedmont: igneous and metamorphic rock
- Valley, Ridge, Plateau: Sedimentary rocks (limestone)

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## Maryland's Five Geologic Areas




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## Major Aquifers in MD

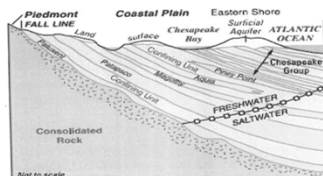


Figure 4. Schematic diagram of the Atlantic Coastal Plain aquifer system (relative thicknesses of aquifers and confining layers not to scale).

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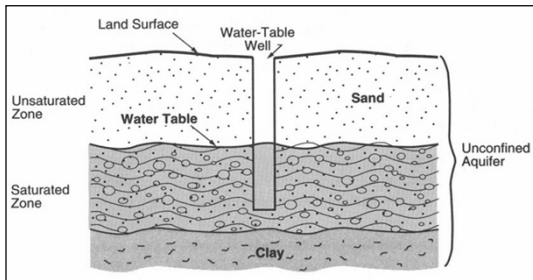
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## Unconfined Aquifer



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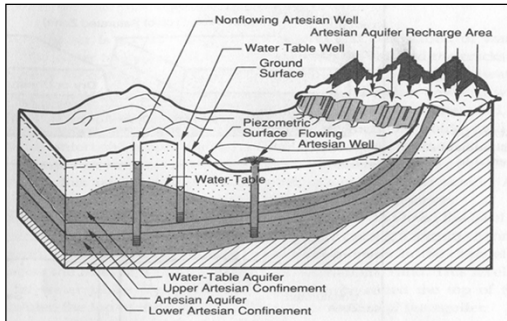
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## What is an Artesian Well?



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## Aquifer Performance...

- Changes in an aquifer are measured by a small diameter test well called an Observation Well
- Located near an operating well

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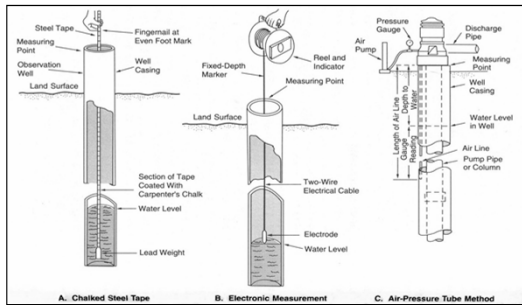
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## Aquifer Evaluation can be Measured Manually or Mechanically



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## Well Yield- Rate of water withdrawal that a well can supply over a long period of time

- Example: Pumpage from an aquifer continuously exceeds the recharge to the aquifer, draw-down will extend and a safe yield will be reduced

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## Ground Water Under the Direct Influence of Surface Water (GWUDISW)

- Definition: Any water beneath the surface of the ground with significant occurrence of:
  - insects
  - other macroorganisms,
  - algae,
  - or large diameter pathogens such as *Giardia lamblia*
  - OR

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### Ground Water Under the Direct Influence of Surface Water (GWUDISW) [continued]

- Definition: Any water beneath the surface of the ground with significant and relatively rapid shifts in water characteristics such as:
  - turbidity,
  - temperature,
  - conductivity, or
  - pH



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### Groundwater Under Direct Influence of Surface Water

- GWUDISW must meet same treatment technologies as surface water.
- Disinfection is mandatory.
- Filtration is mandatory, unless the system meets the filtration avoidance criteria.

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### Basic Rules for Wellhead Protection

- Restrict access to well
- Inspect protective zones regularly
- Slope ground away from the well
- Locate maintenance sheds and chemical storage outside of protective zone
- Site new wells carefully

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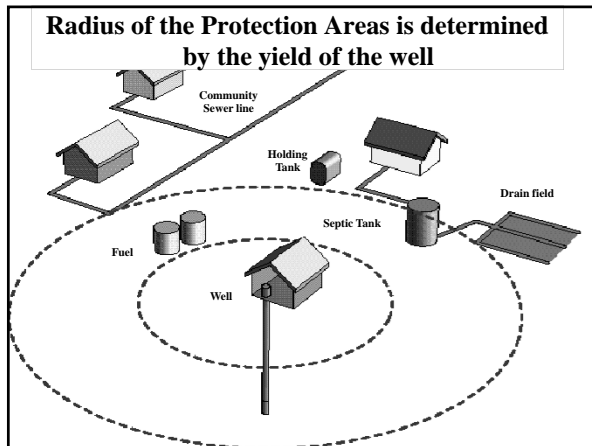
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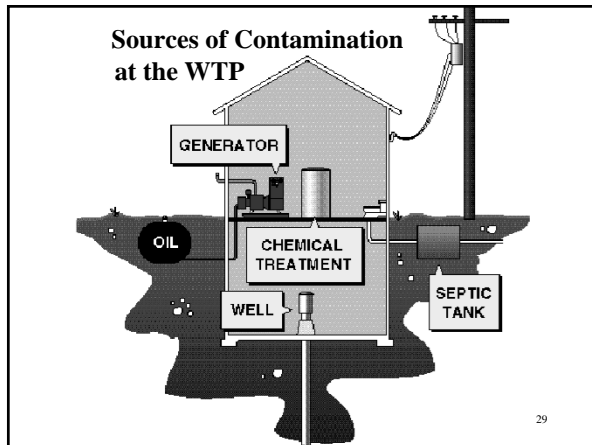
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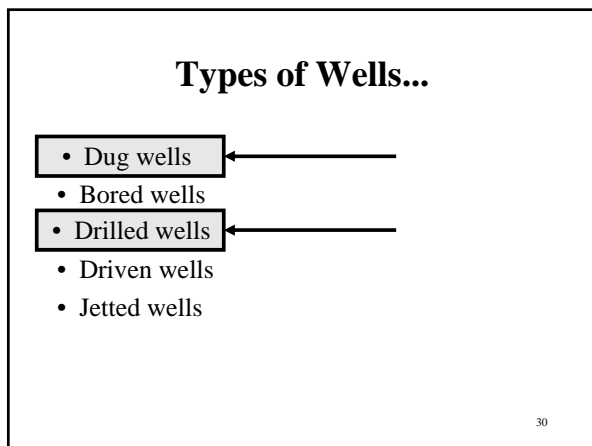
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**Dug Wells  
(found in rural areas)**

- Do not penetrate much below water table
- May fail during drought conditions
- Protection from surface contamination difficult
- Only type of well **always** treated as a surface water source

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**Drilled Wells Most Commonly  
Found In Public Water Supply**

- Benefit - they can reach extreme depths and have large well diameters  
(up to 4 feet and larger)

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**Well Drilling Methods**

- Cable Tool
- Mud Rotary
- Reverse Circulation
- Air Rotary
- Auger
- Other

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## The Well Construction Process

- Project mobilization
- Pilot hole drilling and sample collection
- Geophysical log of pilot hole
- Determination of proper well casing and screen placement
- Verification of proper material placement (casing grout, gravel pack, etc)
- Well development to remove residual materials from the drilling process

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## The Well Construction Process (continued)

- Well and aquifer performance testing to determine important hydraulic parameters
- Water quality testing (physical, chemical, bacteriological)
- Pumping equipment selection, installation, and testing
- Supplemental water treatment

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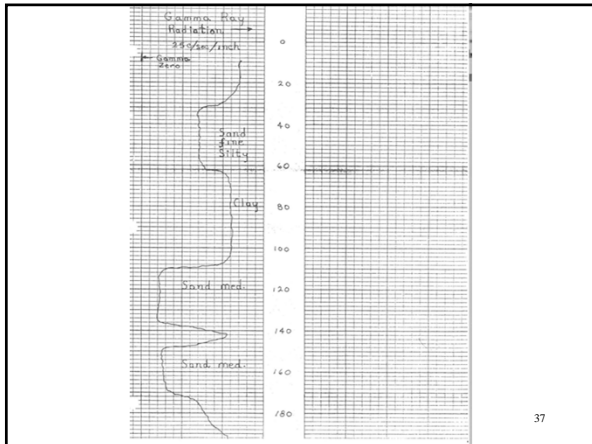
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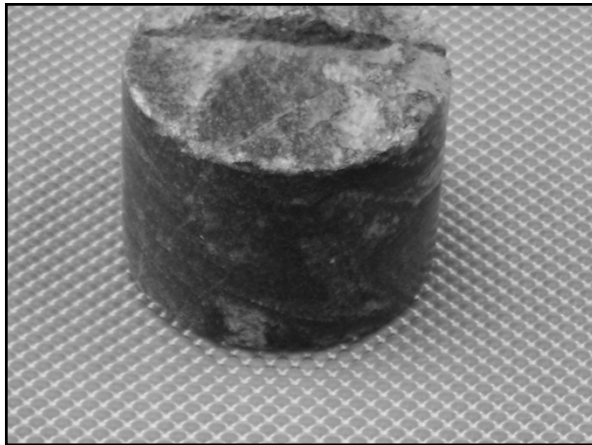
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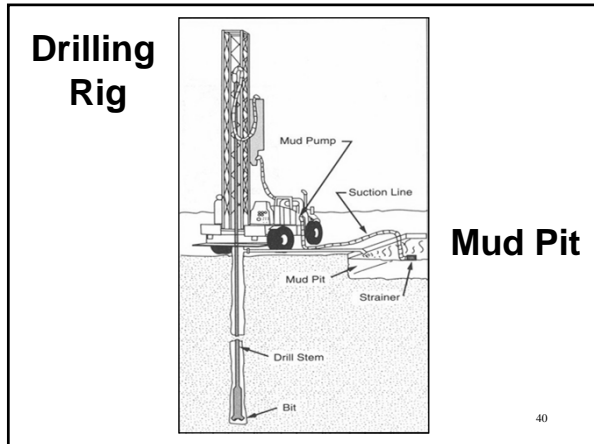
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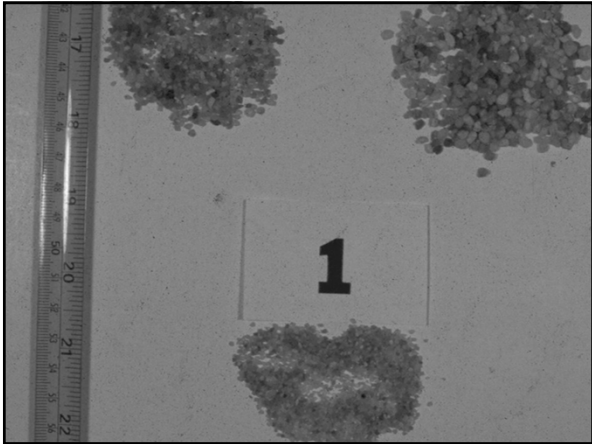
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Well Drilling Gone Wrong!

[www.youtube.com/watch?v=-14FOL\\_UPB0](http://www.youtube.com/watch?v=-14FOL_UPB0)

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## Sampling Procedures for New Wells

- Sampling and testing required prior to placing a well in operation
- Use state-certified lab and approved methods
- Well must be fully-developed prior to sampling

46

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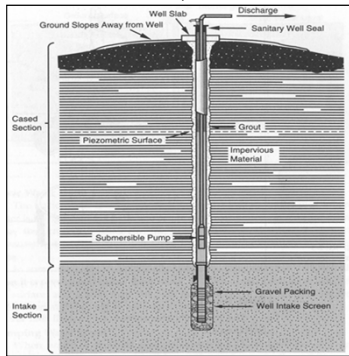
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## Highlight Well Casing, Well Slab, Well Grout, Well Screen



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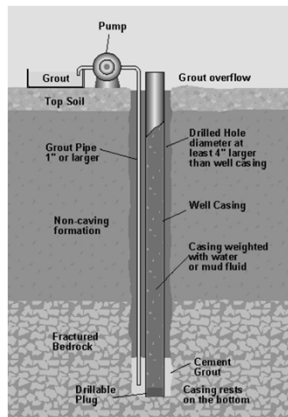
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## Ground Water Wells: Grouting Methods



Well casing is cemented by pumping grout through a pipe lowered into the annular space outside the casing.

Source: Ground Water and Wells, Johnson Well Screen Division, 1975

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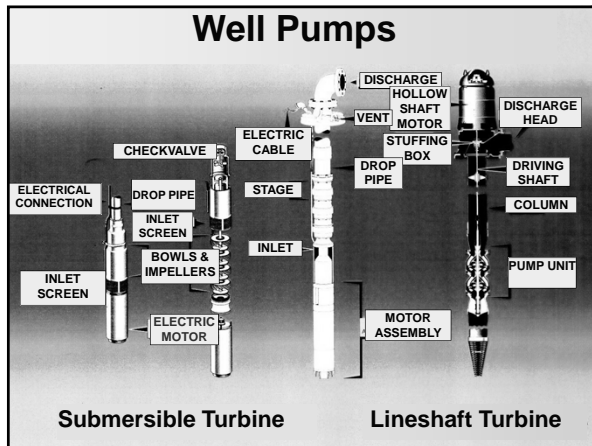
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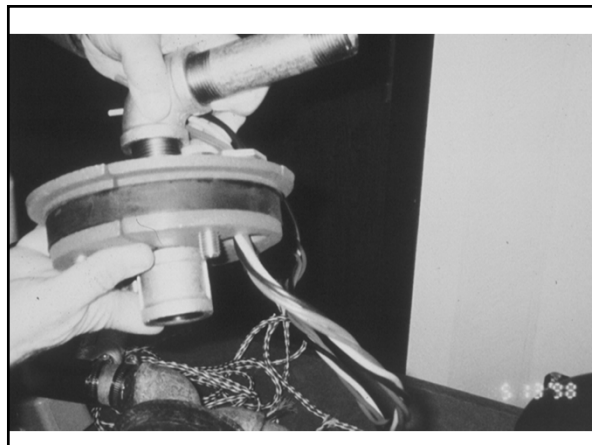
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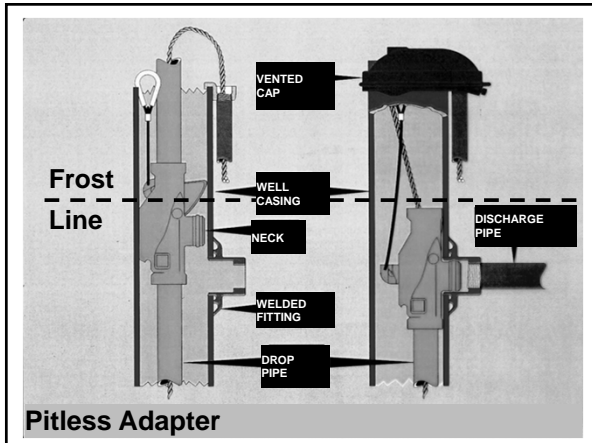
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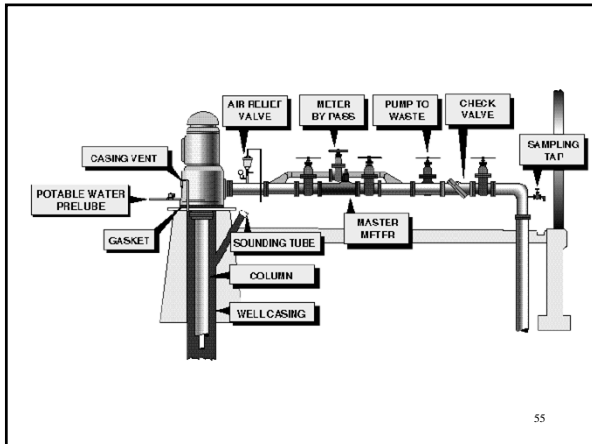
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## Why are some inner casings not brought to the surface?

- Advantages
  - \$
  - More room for the pump
- Disadvantages
  - Access
  - Maintenance and repairs
- Relief screen
  - Expensive lessons

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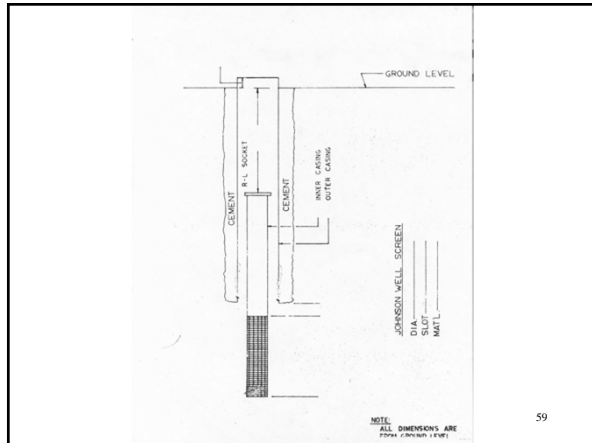
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## Well Operation & Maintenance

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## O&M

- Measure Drawdown
- Calculate Specific Capacity
  - GPM per Foot of Drawdown
- Check for presence of sand
- Check actual pump capacity
- Verify pressure cut-in and cut-out set points
- Ensure that connections & sanitary seals are intact
- Eliminate potential sources of contamination

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## O&M

- Test alarm systems
- Record daily water production and pump runtime
- Measure raw water quality
  - Watch for changes in contaminant levels
- Consult with a “Well System Professional”
  - Well Driller
  - Hydrogeologic / Engineering Firm
  - Electricians
    - Power
    - Industrial Controls

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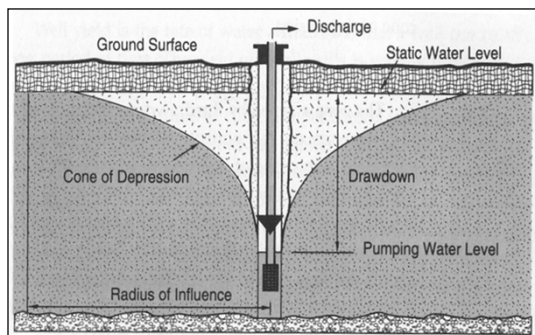
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## Draw-down



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## HOW/WHY DO WELLS CLOG UP???

- PHYSICAL PLUGGING
- CHEMICAL / MINERAL ENCRUSTATION
- BIOLOGICAL GROWTHS
- COMBINATIONS
- VELOCITY CONSIDERATIONS

64

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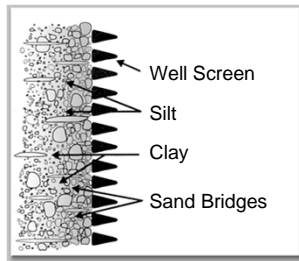
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## Cause: Physical Plugging

Physical and mechanical blockage that occurs when fine-ground material migrates toward the well bore accumulating in pore spaces near the well screen, gravel pack, bore hole wall, or in rock wells in fracture apertures.



Orange juice pulp

65

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## Mineral Encrustation - Type 1



Red or brown deposits

66

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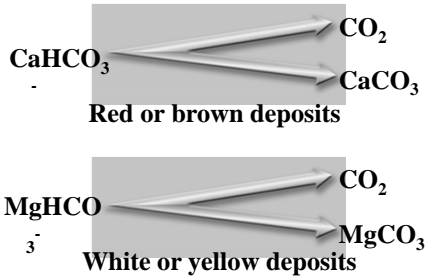
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### Mineral Encrustation - Type 3



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### Biological Clogging

- Iron-related bacteria (IRB)
  - Stalks (gallionella)
  - Sheaths (crenothrix)
  - Filamentous
- Sulfate-reducing bacteria
  - Microbial corrosion
  - Hydrogen sulfide gas
  - Rotten-egg odor
- Slime-forming bacteria
  - Mostly IRBs

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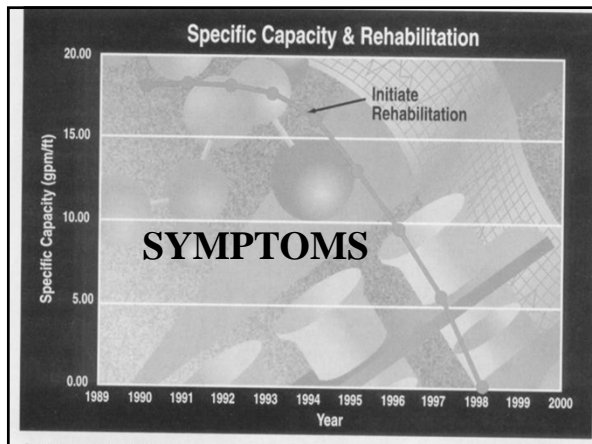
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**SO HOW DO WE FIX IT ???**

- WELL REDEVELOPMENT OR WELL REHABILITATION
- UNCLOG OR REOPEN THE PASSAGEWAYS
- LET THE WATER FLOW INTO THE WELL EASIER (LIKE IT USED TO)
- 

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**WELL REHABILITATION**

- LIKE CLEANING YOUR TEETH
- BRUSHES
- CHEMICALS
- BIOFILMS
- EVERYONE HAS A FAVORITE

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**Types of Chemical Treatment**

- Chlorine/ Disinfectants as Biocides
- Acids to Dissolve Encrustation
- Formulated chemistries
- Aqua Freed™
- **Not Focused**
- **Disposal Considerations**

72

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## Mechanical Redevelopment Techniques

- Over-pumping/surging
- Wire-brushing
- Solid swab
- Double (hollow) swab
- High-velocity jetting
- Hydraulic fracturing
- Air pressurization
- Sonar jetting (primer cord)
- Aqua-burst

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### Surge Block



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### Schultes Ultra High Pressure

- 5000 to 10,000 Psi
- Cleans casings and screens
- Removes encrustation and opens fractures
- Exposes weak Joints or other Problems
- Powerful
- Focused

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### Jetting Tools



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## 10,000 PSI JETTING



Before cleaning



After cleaning

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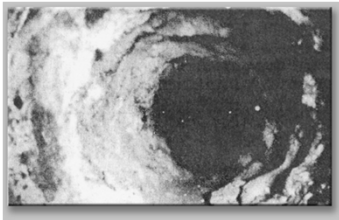
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## Video Log



GOOD DIAGNOSTIC TOOL

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## Jetting Video

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## Contamination...

Any microorganisms, chemicals, wastes or wastewater in a concentration that makes the water unfit for its intended use

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Anything more than:  
2 Hydrogens & 1 Oxygen  
( $H_2O$ ), the water is contaminated.

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What if you add another oxygen to  
 $H_2O$  ?

You would have:  
 $H_2O_2$   
Hydrogen Peroxide

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### Contamination May Come From Natural Pollutants...

- Turbidity
- Radium 226 & 228
- Total Coliforms

86

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### Or Manmade Chemicals...

- Synthetic organic compounds
- Volatile organic compounds

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What is the main determining factor when deciding on the type of treatment needed ?

- The characteristics of the raw water source (the contaminants that must be removed)

88

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### Public Systems Using Groundwater...

- Previously, many systems provided acceptable quality without providing treatment
- Many more systems have turned to treatment for the following reasons:
  - Meet federal and state requirements for disinfection
  - Remove contaminants posing a threat to public health
  - Remove contaminants that reduce the aesthetic quality of the water

89

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### Groundwater can contain...

- High hardness
- Objectionable contaminants:
  - Iron
  - Manganese
  - Hydrogen Sulfide
- Radionuclides
- Synthetic contaminants:
  - Pesticides
  - Herbicides
  - Industrial solvents

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## Groundwater Treatment Options

- All treatment must be approved by the state regulatory agency.



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## GW Treatment Options

### Corrosion Control:

To prevent corrosion in the distribution system and prevent leaching of lead & copper in household plumbing.

### Methods:

- Feeding a pH adjustment chemical
- Feeding a corrosion inhibitor chemical

92

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## GW Treatment Options

### Iron / Manganese Removal:

To prevent iron / manganese from...

- forming deposits inside pipes and fixtures
- staining laundry
- causing taste and odor complaints

### Methods:

- Feeding a chemical to sequester the iron / manganese
- Physical removal (oxidation / filtration)
- Ion Exchange

93

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## GW Treatment Options

### Calcium / Magnesium (Hardness) Removal:

To prevent calcium / magnesium from...

- forming deposits inside pipes and fixtures
- causing customer complaints
  - Hard water will not allow soap bubbles to form

### Methods:

- Ion Exchange
- Lime / Soda Ash Softening

94

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## Groundwater Treatment Options

### Disinfection:

To prevent microorganisms from entering/forming in the distribution system.

### Methods:

- Application of a disinfecting chemical
  - Chlorine
  - Sodium Hypochlorite
  - Calcium Hypochlorite
- Ultraviolet (UV) light

95

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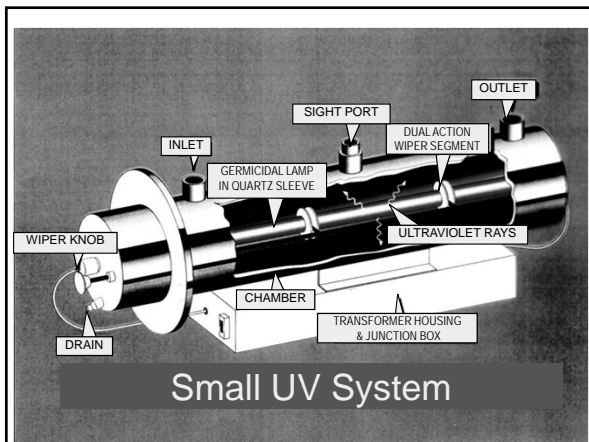
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## Groundwater Treatment Options

### Organics Removal:

To prevent volatile and synthetic organics from entering the distribution system and causing chronic health problems.

### Methods:

- Aeration
- GAC Contactors

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## Groundwater Treatment Options

### Arsenic Removal:

To prevent arsenic from entering the distribution system and causing chronic health problems.

### Methods:

- Membrane Filtration
- Ion Exchange
- Lime Softening
- Oxidation / Filtration
  - May also incorporate coagulation

98

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## In Situ Treatment...

- Chlorine, introduced into a well, is a frequently used method of removing iron bacteria from a well.

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# Water Treatment Chemicals

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## Disinfection Chemicals

- Chlorine
- Calcium Hypochlorite
- Sodium Hypochlorite
- Ozone
- Ultraviolet

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## Chemicals Used to Prevent Corrosion...

- Calcium Hydroxide....(lime)
- Calcium Oxide.....(quicklime)
- Sodium Hydroxide.....(caustic soda)
- Sodium Carbonate.....(soda ash)
- Sodium Bicarbonate....(baking soda)
- Phosphates

102

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## Chemicals Used to Prevent Scaling...

- Carbon Dioxide
- Sulfuric Acid
- Phosphates

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## All Chemicals Added to Drinking Water

- MUST be approved by the National Sanitation Foundation (NSF)



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## Chemical Feed Systems at Small Water Systems

- Gas Feed Systems
- Liquid Chemical Feed Systems
- Dry Chemical Feed Systems

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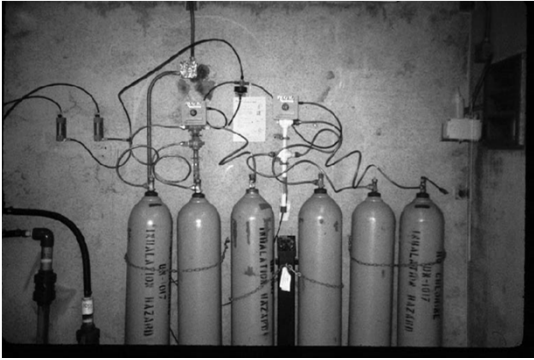
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## Pressurized System (pigtailed)




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## Simple vacuum controlled chlorinator

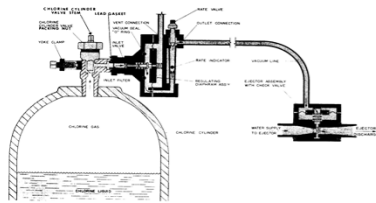


Fig. 7.16 Direct cylinder mounted connection from chlorine gas supply to chlorinator  
(Permission of Capital Controls Company, Canby, OR)

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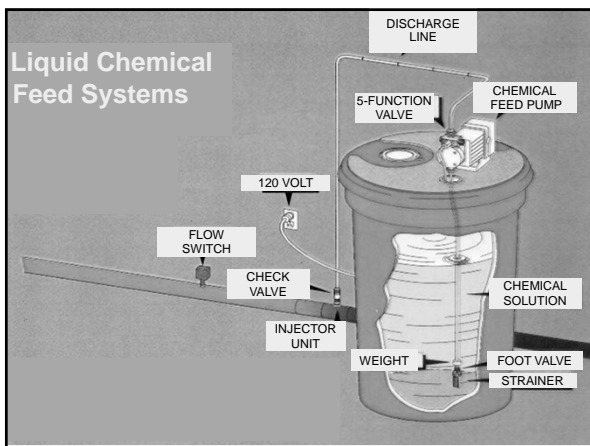
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## Liquid Chemical Feed Systems




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## Diaphragm-Type Metering Pump



- Speed Control
- Manual / Flow Pace Selector
- Stroke Control

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## Peristaltic Pump



- Speed Control

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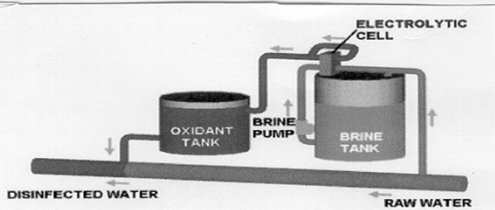
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## Sodium Hypochlorite Produced On-site



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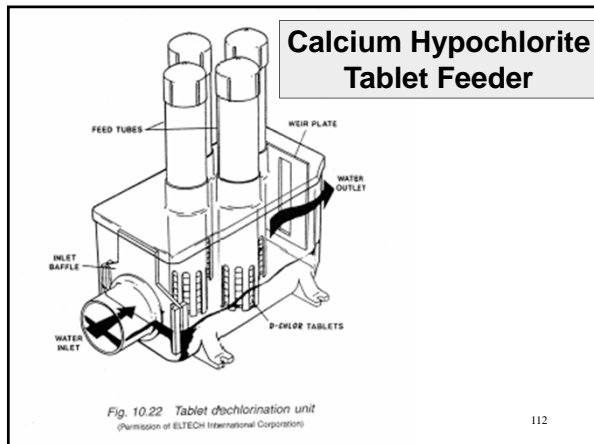
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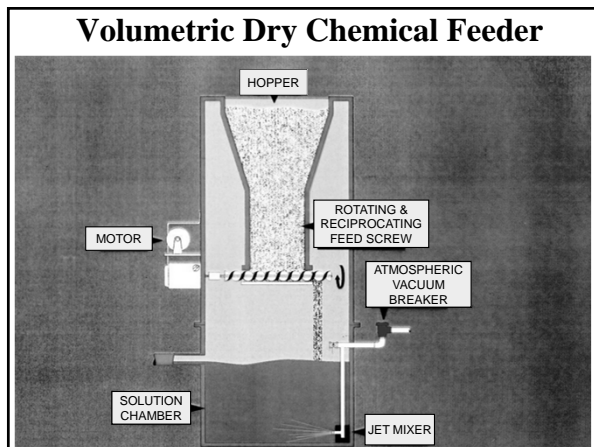
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- ### Treatment Processes
- Disinfection
  - Corrosion / Scaling Control
  - Iron & Manganese Removal
  - Organics Removal
  - Arsenic Removal
- 114

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## Disinfection !

**The one process that we can't do without.**

Gastroenteritis  
Typhoid  
Dysentery  
Cholera  
Hepatitis

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Coliform Bacteria is tested for by the presence or absence method....

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## Important Terms

Dosage : The amount of chlorine added mg/l or ppm

Demand : The amount of chlorine required to react with the organic and inorganic substances.

Residual : Dose ( - ) Demand = Residual  
The amount remaining after contact time.

Free Residual : Exists as Hypochlorous acid or hypochlorite

Combined Residual : Chlorine which has combined with ammonia to form Chloramines.

Total Residual : Is the sum of free and combined residual

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## Factors Influencing Disinfection

- pH
- Temperature
- Turbidity
- Organic Matter
- Inorganic Matter
- Reducing Agents
  - Soluble iron & manganese
  - Nitrite
  - Hydrogen sulfide
- Microorganisms

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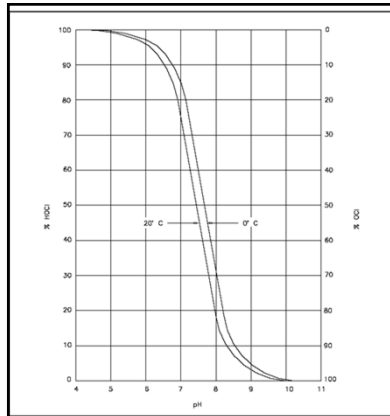


Figure 1 - Distribution of HOCl and OCl- In Water at Various pH Levels

119

## Effects of pH

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## Why Corrosion and Scaling Control?

- Protect public health
- Improve water quality
- Extend life of plumbing equipment
- Meet state and federal regulations

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### Stabilization...

- Controlling of damaging corrosion or deposit scaling on pipelines due to source water

121

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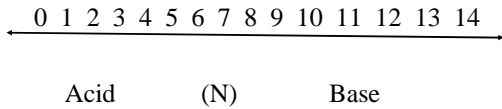
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### pH Scale



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### Protecting Public Health...

- Corrosion can raise toxic metal levels in the water
  - example - lead and copper
- Corrosion can cause tubercles
  - Tubercles can protect bacteria from disinfection

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### Meeting Regulations...

- Meeting Lead and Copper regulations enacted in 1991
- Corrosion control when levels reach Action Levels

124

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### Water Quality...

- **Taste** - Metallic taste from corrosion of copper
- **Odor** - Dissolved iron from corrosion acts as a food source for iron bacteria which can cause taste and odor problems
- **Color** - Corrosion can cause stains such as red, blue or green

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### Corrosion Can be in Two Categories...

- Localized
- Uniform

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## Scale Formation...

- Small amount of scale on the inside of pipes can protect against corrosion
- Too much scale can reduce the carrying capacity of pipe or system

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## What is Hard Water?

- Water having high concentrations of calcium and magnesium ions

0-60 mg/l  $\text{CaCO}_3$  – Soft Water  
61-120 mg/l  $\text{CaCO}_3$  – Moderately Hard Water  
121-180 mg/l  $\text{CaCO}_3$  – Hard Water  
>180 mg/l  $\text{CaCO}_3$  – Very Hard Water

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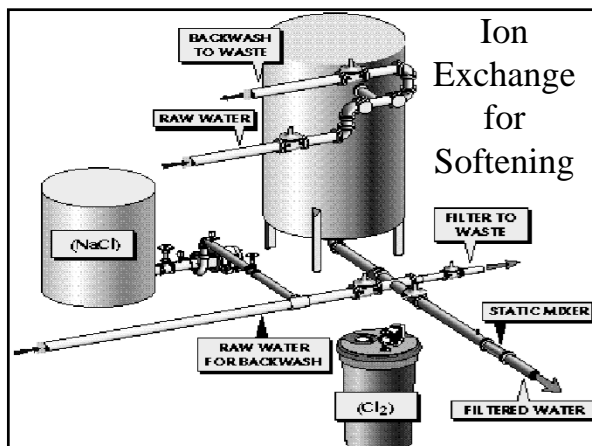
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### The Problem with Iron

- Iron exists in groundwater in ferrous iron and is in the soluble form ( $\text{Fe}^{2+}$ )
- When soluble iron comes into contact with oxygen (or oxidizing compound), insoluble precipitate (ferric iron) is formed ( $\text{Fe}^{3+}$ )
- Ferric iron stains plumbing fixtures, laundry, etc.
- Iron supports the growth of iron reducing bacteria which causes a biofilm (slime) to form inside pipes.
- Also, iron can cause encrustation inside water mains

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### **Iron and Manganese Removal**

- Iron and manganese in surface water and ground water
- Treatment Processes
  - Oxidation and filtration
  - Oxidation, clarification, and filtration
  - Manganese greensand filtration
  - Ion exchange
  - Sequestering

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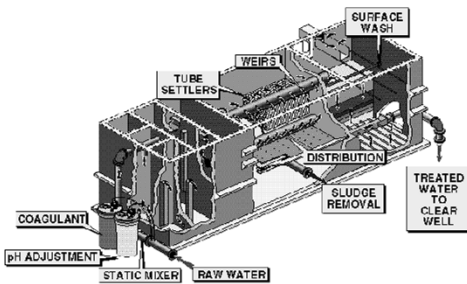
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### **Package Treatment Plants**



Conventional Treatment - Package Plant

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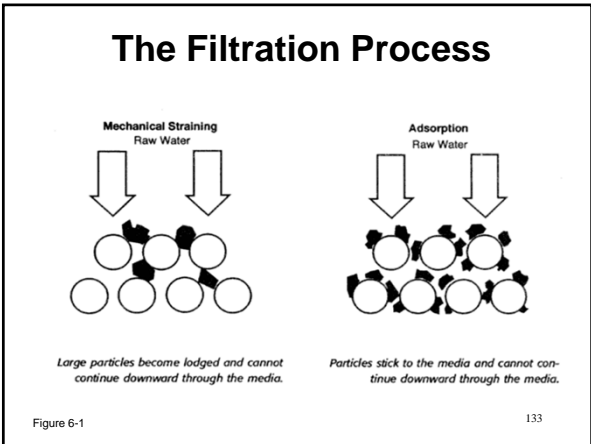
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## Arsenic Removal

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- ### Regulatory Issues
- SDWA identifies Arsenic as a Primary Contaminant
  - Chronic health effects
  - MCL
    - 10 parts per billion (0.01 mg/L)
- 135

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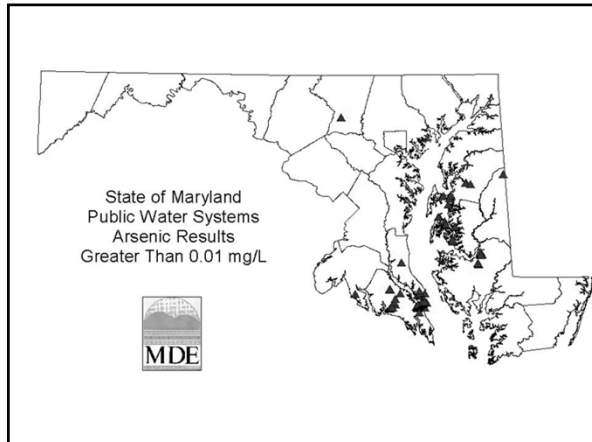
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### Arsenic Removal Processes

- Ion Exchange
- Reverse Osmosis / Nanofiltration
- Lime Softening
- Oxidation / Filtration
  - Conversion of As (III) to As (V)
  - Conventional Fe / Mn treatment systems are effective
  - Activated Alumina for coagulation

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  - Small Water System O & M
  - <http://www.owp.csus.edu/courses/drinking-water.php>
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  - Water Supply Operations
  - Water Treatment
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